The Baylis – Hillman Adducts as Valuable Source for One-Pot Multi-Step Synthesis: A Facile Synthesis of Substituted Piperidin-2-ones

by Deevi Basavaiah*, Raju Jannapu Reddy, and Dandamudi V. Lenin

School of Chemistry, University of Hyderabad, Hyderabad-500 046, India $(fax: +91-40-23012460; e-mail: dbsc@uohyd.ernet.in)$

A facile, convenient, and one-pot multi-step synthesis of substituted piperidin-2-ones from the Baylis – Hillman alcohols derived from various aldehydes and acrylonitrile, involving Johnson – Claisen rearrangement, reduction of an α , β -unsaturated nitrile moiety into the saturated amine-skeleton, followed by cyclization, in an operationally simple procedure, is described.

Introduction. – In continuation of our interest in the synthesis of heterocyclic molecules [1], we herein report a simple, facile, and one-pot multi-step synthesis of piperidin-2-one derivatives, that is, 5-substituted and 3,5-disubstituted piperidin-2-ones from Baylis-Hillman adducts. The piperidin-2-one moiety occupies a special place in nitrogen heterocyclic chemistry because of the presence of this moiety in a variety of biologically important compounds and therefore, the development of simple procedures for the synthesis of such frameworks is an attractive endeavor in heterocyclic chemistry $[2]$ ¹).

Results and Discussion. – Applications of the Baylis – Hillman adducts in various organic transformation methodologies have been well documented in the literature [1] [4] [5]. We have, some time ago, developed a stereo-defined synthesis of functionalized trisubstituted alkenes, *i.e.*, ethyl $(4Z)$ -4-cyanoalk-4-enoates *via* the Johnson – Claisen (Claisen orthoester) rearrangement²) of the Baylis – Hillman alcohols, which represent CN-substituted allylic alcohols, with triethyl orthoacetate (*Scheme 1*) [7a]³).

Scheme 1. Johnson – Claisen Rearrangement of 3-Hydroxy-2-methylidenealkanenitriles [6]

¹⁾ For a leading review on synthesis of piperidones and piperidines, see [3].

²) For leading references on the *Johnson – Claisen* rearrangement, see [6].

³) For other *Johnson – Claisen* rearrangement of *Baylis – Hillman* adducts, see [7b] [7c].

^{© 2010} Verlag Helvetica Chimica Acta AG, Zürich

It occurred to us that (4Z)-4-cyanoalk-4-enoates would be excellent starting materials for the synthesis of piperidin-2-one derivatives by complete reduction of the ene-nitrile functionality into an amino group, which would be in the proximity of the ester group for a possible cyclization to a six-membered heterocyclic framework with a provision for having a different substitution profile at the 3- and 5-positions. It also occurred to us that the entire operation can be performed in one-pot, starting from the Baylis – Hillman adducts involving a Johnson – Claisen rearrangement and reduction, followed by cyclization, to provide the piperidin-2-one framework (*Scheme 2*).

Accordingly, we have first selected 2-[hydroxy(phenyl)methyl]prop-2-enenitrile (1a), a Baylis-Hillman alcohol $(B-H \text{ alcohol})$ derived from benzaldehyde and acrylonitrile $(=$ prop-2-enenitrile), as a substrate for this multi-step probing. The best result was obtained when 1a (1 mmol) was treated with triethyl orthoacetate (1 ml) at 145° in the presence of a catalytic amount of propanoic acid (3 drops) for 2 h, followed (after removing the excess of orthoester under reduced pressure) by treatment with CoCl₂ \cdot 6 H₂O (2 mmol)/NaBH₄ (10 mmol)⁴) in MeOH at 0° for 30 min and then for 6 h at room temperature, thus providing 5-benzylpiperidin-2-one (2) as a colorless solid in 60% yield, after workup and purification by column chromatography (Scheme 3 and Table 1, Entry 1). This was indeed an encouraging result in the sense that three steps were performed in one-pot to produce the desired 2 in an acceptable yield.

Scheme 3. One-Pot Three-Step Synthesis of 5-Substituted Piperidin-2-ones

To understand the general character of this reaction, we have, according to *Scheme 3*, successfully transformed representative $B - H$ alcohols **1b** – **1f**, obtained from various aromatic aldehydes and acrylonitrile, into the corresponding 5-arylmethylpi-

⁴⁾ For literature reports on reduction of CN group using $CoCl₂ · 6 H₂O/NaBH₄$, see [8].

Entry	$B-H$ alcohol	R	\mathbb{R}^1	Product ^b)	Yield ^c) [%]
	1a	C_6H_5	Н		60
2	1b	$2-MeC_6H_4$	Н		58
3 ^d	1c	$4-MeC6H4$	Н		64
$\overline{4}$	1d	$4-(Pr)C_6H_4$	Н		56
5	1e	2 -ClC ₆ H ₄	Н	o	53
6	1f	$4-CIC6H4$	Н		55
7 ^d	1g	$C_6H_5CH_2CH_2$	Н	8	59
8	1h	Heptyl	Н	q	65
9^e	1a	C_6H_5	Me	10	61

Table 1. One-Pot Three-Step Synthesis of Substituted piperidin-2-ones^a)

^a) All reactions were carried out on a 1 mmol scale of $B-H$ alcohols **1a**-**1h**, which were prepared according to a literature procedure [9]. $\frac{b}{c}$ Compounds 2–10 were obtained as colorless solids and were fully characterized (see *Exper. Part*). ^c) Yields are of the pure products, based on $B-H$ alcohols. ^d) The structures of these molecules were also established by X-ray diffraction analysis (*Figs. 1* and 2, *Table* 2). e) Isolated as a 1 : 1 mixture of diastereoisomers.

peridin-2-ones $3 - 7$ in 53-64% yield of isolated products (*Table 1, Entries 2-6*). With the same procedure, we also have successfully transformed $B - H$ alcohols 1g and 1h, obtained from the corresponding aliphatic aldehydes (hydrocinnamaldehyde and octanal) and acrylonitrile, into 5-alkylpiperidin-2-ones 8 and 9 in 59 and 65% yield, respectively (Table 1, Entries 7 and 8). Piperidin-2-ones 4 and 8 delivered suitable single crystals for an X-ray crystal-structure analysis, which further confirmed the structures of these compounds (*Figs. 1* and 2, *Table* 2).

Fig. 1. ORTEP Diagram of compound 4 (H-atoms are omitted for clarity)

Fig. 2. ORTEP Diagram of compound 8 (H-atoms are omitted for clarity)

	$\overline{\mathbf{4}}$	8
Crystallized from	MeOH	MeOH
Empirical formula	$C_{13}H_{17}NO$	$C_{14}H_{19}NO$
Formula weight $\lceil g \text{ mol}^{-1} \rceil$	203.28	217.30
Crystal color, habit	colorless, plate	colorless, block
Crystal dimensions [mm]	$0.42 \times 0.21 \times 0.07$	$0.46 \times 0.38 \times 0.23$
Temperature $[K]$	298	298
Crystal system	Monoclinic	Triclinic
Space group	P2 ₁ /c	$P\bar{1}$
Z	$\overline{4}$	2
Reflections for cell determination	2316	2438
20 Range for cell determination $\lceil \cdot \rceil$	$1.43 - 26.27$	$1.49 - 26.07$
Unit cell parameters a [A]	14.335(5)	5.8744(15)
$b\vert\text{A}\vert$	5.812(2)	8.040(2)
$c [\AA]$	14.026(5)	14.260(4)
α [\degree]	90.00	74.375(4)
β [\degree]	96.357(6)	89.430(4)
γ [$^{\circ}$]	90.00	73.780(4)
$V[\AA^3]$	1161.4(7)	621.2(3)
F(000)	440	236
D_x [g cm ⁻³]	1.163	1.162
μ (Mo K_{α}) [mm ⁻¹]	0.073	0.072
Transmission factors [min; max]	0.9700; 0.9949	0.9674; 0.9835
Scan type	Φ and ω	Φ and ω
$2\theta_{\text{max}}$ [°]	26.27	26.07
Reflections measured	11271	6507
Symmetry-independent reflections	2316	2438
$R_{\rm int}$	0.0270	0.0244
R_1, wR_2 [$I \geq 2\sigma$ (1)]	0.0464, 0.1246	0.0571, 0.1589
R_1 , w R_2 (all data)	0.0684, 0.1386	0.0716, 0.1714
Goodness-of-fit on F^2	1.045	1.023

Table 2. Crystallographic Data for Compounds 4 and 8^a)

With the intention to extend this new synthesis to the preparation of 3,5disubstituted piperidin-2-ones, we have used triethyl orthopropanoate for the Johnson – Claisen rearrangement. Thus, $B - H$ alcohol 1a, on successive treatment with triethyl orthopropanoate and $CoCl₂ · 6 H₂O/NaBH₄$ according to the same procedure (as in the case of triethyl orthoacetate) provided 5-benzyl-3-methylpiperidin-2-one (10) in 61% yield (Scheme 3 and Table 1, Entry 9). However, the ¹H- and ¹³C-NMR spectra clearly indicate that this molecule is $ca. 1:1$ mixture of diastereoisomers. In the ¹H-NMR spectrum, $Me - C(3)$ appeared as two *doublets* of equal intensities at 1.19 and 1.23 ppm, thus indicating for 10 a 1:1 mixture of diastereoisomers. Two singlets at 6.46 and 6.50 ppm are attributed to the NH groups of the diastereoisomers. In the 13C-NMR spectrum, two signals at 175.7 and 176.2 ppm for one $C=O$, two signals at 139.2 and

^a) Detailed X-ray crystallographic data of 4 and 8 are available from the *Cambridge Crystallographic* Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK under CCDC-635054 and -635055, respectively.

139.4 ppm for one quaternary C-atom of the Ph group, and twelve signals for six aliphatic C-atoms clearly indicate that this compound is a mixture of diastereoisomers.

In conclusion, we have successfully developed a convenient and operationally simple one-pot three-step procedure for the synthesis of 5-substituted and 3,5 disubstituted piperidin-2-ones from Baylis – Hillman alcohols, by Johnson – Claisen rearrangement, reduction, and cyclization, thus demonstrating the application of Baylis – Hillman adducts as a versatile source for one-pot multi-step syntheses.

Experimental Part

General. M.p.: Superfit (India) capillary melting point apparatus; uncorrected. IR Spectra: JASCO- FT -IR model 5300 spectrometer using solid samples as KBr plates, in cm⁻¹. ¹H- (400 MHz) and ¹³C-NMR (50 MHz) spectra: in CDCl₃ on a *BRUKER-AVANCE-400* and *BRUKER-AC-200* spectrometer; δ in ppm, J in Hz; tetramethylsilane (TMS, 0 ppm) as internal standard. MS: Shimadzu-LCMS-2010A mass spectrometer. Elemental analyses: Thermo-Finnigan Flash EA 1112 analyzer. X-Ray diffraction measurements: at 298 K on a Bruker SMART APEX CCD area detector system, using graphite monochromated Mo K_a radiation and fine-focus sealed tube ($\lambda = 0.71073 \text{ Å}$).

5-Benzylpiperidin-2-one (2). To a stirred soln. of propanenitrile 1a $(1 \text{ mmol}, 0.159 \text{ g})$ in triethyl orthoacetate ($=1,1,1$ -triethoxyethane; 1 ml) was heated at 145° in the presence of a cat. amount of propanoic acid (3 drops) for 2 h. Excess orthoester (along with propanoic acid) was distilled off under reduced pressure. The residue was diluted with MeOH (8 ml) , and CoCl₂ \cdot 6 H₂O $(2 \text{ mmol}, 0.476 \text{ g})$ was added. The resulting soln. was cooled to 0° , and NaBH₄ (10 mmol, 0.38 g) was added in three portions within 15 min (H_2) gas evolution was observed). The mixture (black formed precipitate) was stirred for 30 min at 0° , and then allowed to warm to r.t. and stirred for further 6 h. MeOH was removed under reduced pressure. Then, the residue was diluted with 4N HCl (15 ml) and extracted with AcOEt (3 \times 20 ml). The combined org. layers were washed with H₂O (20 ml) and dried over anh. Na₂SO₄. After evaporation of the solvent, the residue was purified by CC (SiO₂; AcOEt) to furnish 5-benzylpiperidin-2one (2) as a colorless solid in 60% (0.114 g) yield. M.p. 107 – 109°. IR: 3300 – 2800 (multiple bands), 1662. 1 H-NMR: 1.46 – 1.60 (m, 1 H); 1.85 – 1.96 (m, 1 H); 2.00 – 2.15 (m, 1 H); 2.25 – 2.49 (m, 2 H); 2.55 – 2.70 $(m, 2 H)$; 2.95 – 3.04 $(m, 1 H)$; 3.20 – 3.29 $(m, 1 H)$; 6.27 (br. s, 1 H, D₂O exchangeable); 7.14 $(d, J = 6.8,$ 2 H); 7.19 – 7.34 (m, 3 H). 13C-NMR: 26.9; 30.7; 35.1; 39.6; 47.3; 126.5; 128.6; 128.9; 139.2; 172.6. LC-MS: 190 ($[M+H]^+$). Anal. calc. for C₁₂H₁₅NO: C 76.16, H 7.99, N 7.40; found: C 76.18, H 8.04, N 7.45.

5-(2-Methylbenzyl)piperidin-2-one (3). Yield: 58%. M.p. 103 – 105°. IR: 3300 – 2800 (multiple bands), 1666. ¹H-NMR: 1.49 – 1.64 $(m, 1 H)$; 1.83 – 1.97 $(m, 1 H)$; 2.00 – 2.12 $(m, 1 H)$; 2.25 – 2.50 (m, m) $5 H$); $2.63 (d, J = 7.2, 2 H)$; $2.97 - 3.07 (m, 1 H)$; $3.21 - 3.30 (m, 1 H)$; $6.35 (br. s, 1 H, D₂O exchangeable)$; 7.05 – 7.20 (m, 4 H). 13C-NMR: 19.4; 27.0; 30.6; 33.9; 36.7; 47.2; 125.9; 126.5; 129.7; 130.5; 135.9; 137.5; 172.7. LC-MS: 204 ($[M + H]^+$). Anal. calc. for C₁₃H₁₇NO: C 76.81, H 8.43, N 6.89; found: C 76.67, H 8.49, N 6.93.

 $5-(4-Methylbenzyl) piperidin-2-one$ (4). Yield: 64% . M.p. $111-113^\circ$. IR: $3200-2850$ (multiple bands), 1662. ¹H-NMR: 1.45 – 1.58 $(m, 1 H)$; 1.85 – 1.94 $(m, 1 H)$; 1.99 – 2.10 $(m, 1 H)$; 2.25 – 2.49 (m, m) $5 H$); $2.51 - 2.65$ (m, $2 H$); $2.94 - 3.03$ (m, $1 H$); $3.20 - 3.28$ (m, $1 H$); 6.19 (br. s, $1 H$); 7.03 (d, $J = 7.6$, $2 H$); 7.10 (d, J = 7.6, 2 H). ¹³C-NMR: 21.0; 26.8; 30.6; 35.2; 39.2; 47.2; 128.7; 129.2; 135.9; 136.1; 172.6. LC-MS: 204 ($[M+H]^+$). Anal. calc. for C₁₃H₁₇NO: C 76.81, H 8.43, N 6.89; found: C 76.70, H 8.41, N 7.00.

5-[4-(1-Methylethyl)benzyl]piperidin-2-one (5). Yield: 56%. M.p. 95 – 96°. IR: 3320 – 2850 (multiple bands), 1664. ¹H-NMR: 1.23 (d, J = 6.8, 6 H); 1.45 – 1.59 (m, 1 H); 1.86 – 1.95 (m, 1 H); 1.99 – 2.12 (m, 1 H); 2.25 – 2.67 (m, 4 H); 2.83 – 3.04 (m, 2 H); 3.21 – 3.28 (m, 1 H); 6.11 (br. s, 1 H, D₂O exchangeable); 7.06 $(d, J = 7.8, 2 H)$; 7.15 $(d, J = 7.8, 2 H)$. ¹³C-NMR: 24.0; 26.9; 30.6; 33.7; 35.1; 39.2; 47.2; 126.5; 128.8; 136.5; 146.9; 172.7. LC-MS: 232 ($[M + H]^+$). Anal. calc. for $C_{15}H_{21}NO$: C 77.88, H 9.15, N 6.05; found: C 77.96, H 9.17, N 6.03.

 $5-(2-Chlorobenzvl)piperidin-2-one (6)$. Yield: 53% . M.p: $106-108^\circ$. IR: $3200-2950$ (multiple bands), 1672. ¹H-NMR: 1.48 – 1.60 (*m*, 1 H); 1.85 – 1.96 (*m*, 1 H); 2.01 – 2.14 (*m*, 1 H); 2.27 – 2.50 (*m*,

 2 H); $2.54 - 2.70 \text{ (m, 2 H)}$; $2.94 - 3.05 \text{ (m, 1 H)}$; $3.21 - 3.30 \text{ (m, 1 H)}$; 6.05 (br.s, 1 H) ; $7.08 \text{ (d, J = 7.6, 1 H)}$; $7.11 - 7.33$ (m, 2 H); 7.36 (d, $J = 8.0$, 1 H). ¹³C-NMR: 26.9; 30.6; 33.7; 37.0; 47.2; 126.8; 128.0; 129.8; 131.1; 134.2; 137.1; 172.5. LC-MS: 224 ($[M + H]^+$), 226 ($[M + 2 + H]^+$). Anal. calc. for C₁₂H₁₄ClNO: C 64.43, H 6.31, N 6.26; found: C 64.57, H 6.39, N 6.25.

5-(4-Chlorobenzyl)piperidin-2-one (7). Yield: 55%. M.p: 109-111°. IR: 3300-2950 (multiple bands), 1657. ¹H-NMR: 1.45 – 1.59 (m, 1 H); 1.83 – 1.93 (m, 1 H); 1.98 – 2.12 (m, 1 H); 2.24 – 2.49 (m, 2 H); 2.52 – 2.68 (m, 2 H); 2.92 – 3.03 (m, 1 H); 3.18 – 3.29 (m, 1 H); 6.15 (br. s, 1 H); 7.06 (d, $J = 8.0$, 2 H); 7.27 (d, J = 8.0, 2 H). ¹³C-NMR: 26.7; 30.5; 35.0; 38.8; 47.1; 128.6; 130.2; 132.2; 137.7; 172.5. LC-MS: 224 ($[M + H]^+$), 226 ($[M + 2 + H]^+$). Anal. calc. for C₁₂H₁₄ClNO: C 64.43, H 6.31, N 6.26; found: C 64.28, H 6.30, N 6.21.

 $5-(3-Phenylpropyl) piperidin-2-one$ (8). Yield: 59%. M.p. $88-90^\circ$. IR: 3300 – 2840 (multiple bands), 1664. ¹ H-NMR: 1.29 – 1.50 (m, 3 H); 1.56 – 1.95 (m, 4 H); 2.26 – 2.47 (m, 2 H); 2.56 – 2.66 (m, 2 H); 2.88 – 2.97 (m, 1 H); 3.28 – 3.38 (m, 1 H); 6.06 (br. s, 1 H); 7.13 – 7.31 (m, 5 H). 13C-NMR: 26.9; 28.6; 30.6; 32.6; 33.0; 35.8; 47.4; 125.7; 128.2; 141.9; 172.6. LC-MS: 218 ($[M + H]^+$). Anal. calc. for C₁₄H₁₉NO: C 77.38, H 8.81, N 6.45; found: C 77.55, H 8.81, N 6.48.

5-Octylpiperidin-2-one (9). Yield: 65%. M.p. 59-61°. IR: 3300-2800 (multiple bands), 1674. 1_H -NMR: 0.88 (t, J = 6.8, 3 H); 1.20 – 1.51 (m, 15 H); 1.71 – 1.80 (m, 1 H); 1.84 – 1.95 (m, 1 H); 2.24 – 2.48 $(m, 2 H)$; 2.89 – 2.98 $(m, 1 H)$; 3.29 – 3.38 $(m, 1 H)$; 6.19 (br. s, 1 H, D₂O exchangeable). ¹³C-NMR: 14.1; 22.7; 27.0; 27.1; 29.3; 29.5; 29.7; 30.8; 31.9; 33.1; 47.7; 172.8. LC-MS: 212 ($[M+H]^+$). Anal. calc. for C13H25NO: C 73.88, H 11.92, N 6.63; found: C 73.69, H 11.99, N 6.60.

5-Benzyl-3-methylpiperidin-2-one (10). Yield: 61%. M.p. 84-86°. IR: 3300-2900 (multiple bands), 1658. ¹H-NMR: 1.19, 1.23 (2d, J = 6.8, 3 H); 1.60 – 2.78 (m, 6 H); 2.92 – 3.05 (m, 1 H); 3.15 – 3.30 (m, 1 H); 6.46, 6.50 (2 br. s, NH of diastereoisomers, 1 H); 7.05 – 7.38 (m, 5 H). 13C-NMR: 17.1; 18.5; 31.7; 33.8; 34.3; 35.7; 35.9; 36.3; 39.2; 40.3; 46.9; 47.9; 126.3; 128.5; 128.8; 139.2; 139.4; 175.7; 176.2. ¹ H- and ¹³C-NMR data clearly indicate that this product is a mixture of 1:1 diastereoisomers.

We thank DST (New Delhi) for funding this project. R. J. R. thanks CSIR (New Delhi) for his research fellowship. D. V. L. thanks UGC (New Delhi) for his research fellowship. We thank UGC (New Delhi) for support and providing some instrumental facilities. We thank the National Single Crystal X-Ray Facility funded by DST. We also thank Professor S. Pal, School of Chemistry, University of Hyderabad for helpful discussions regarding X-ray data analysis.

REFERENCES

- [1] D. Basavaiah, D. V. Lenin, B. Devendar, Tetrahedron Lett. 2009, 50, 3538; D. Basavaiah, R. J. Reddy, Org. Biomol. Chem. 2008, 6, 1034; D. Basavaiah, R. J. Reddy, J. Srivardhana Rao, Tetrahedron Lett. 2006, 47, 73; D. Basavaiah, J. Srivardhana Rao, R. J. Reddy, J. Org. Chem. 2004, 69, 7379; D. Basavaiah, D. S. Sharada, A. Veerendhar, Tetrahedron Lett. 2004, 45, 3081; D. Basavaiah, T. Satyanarayana, Chem. Commun. 2004, 32; D. Basavaiah, A. Jaganmohan Rao, Chem. Commun. 2003, 604; D. Basavaiah, A. Jaganmohan Rao, Tetrahedron Lett. 2003, 44, 4365; D. Basavaiah, R. M. Reddy, N. Kumaragurubaran, D. S. Sharada, Tetrahedron 2002, 58, 3693; D. Basavaiah, T. Satyanarayana, Org. Lett. 2001, 3, 3619; D. Basavaiah, M. Bakthadoss, S. Pandiaraju, Chem. Commun. 1998, 1639.
- [2] S. Kumar, C. Flamant-Robin, Q. Wang, A. Chiaroni, N. A. Sasaki, J. Org. Chem. 2005, 70, 5946; D. S. Middleton, A. R. MacKenzie, S. D. Newman, M. Corless, A. Warren, A. P. Marchington, B. Jones, Bioorg. Med. Chem. Lett. 2005, 15, 3957; T. R. Elworthy, E. R. Brill, C. C. Caires, W. Kim, L. K. Lach, J. L. Tracy, S.-S. Chio, Bioorg. Med. Chem. Lett. 2005, 15, 2523; F. Yokokawa, A. Inaizumi, T. Shioiri, Tetrahedron 2005, 61, 1459; H. K. Lee, J. S. Chun, C. S. Pak, Tetrahedron 2003, 59, 6445; A. R. MacKenzie, A. P. Marchington, D. S. Middleton, S. D. Newman, B. C. Jones, J. Med. Chem. 2002, 45, 5365; L. M. Nogle, T. R. Williamson, W. H. Gerwick, J. Nat. Prod. 2001, 64, 716; N. Yamazki, T. Ito, C. Kibayashi, Tetrahedron Lett. 1999, 40, 739; R. Bonjouklian, T. A. Smitka, A. H. Hunt, J. L. Occolowitz, T. J. Perun Jr., L. Doolin, S. Stevenson, L. Knauss, R. Wijayaratne, S. Szewczyk, G. M. L. Patterson, Tetrahedron 1996, 52, 395.
- [3] P. M. Weintraub, J. S. Sabol, J. M. Kane, D. R. Borcherding, Tetrahedron 2003, 59, 2953; K. Takasu, N. Nishida, M. Ihara, Synlett 2004, 1844; R. G. Arrayás, A. Alcudia, L. S. Liebeskind, Org. Lett. 2001, 3, 3381; E. Jao, P. B. Slifer, R. Lalancette, S. S. Hall, J. Org. Chem. 1996, 61, 2865; T. Fujji, S. Yoshifuji, Tetrahedron 1970, 26, 5953.
- [4] C. E. Aroyan, A. Dermenci, S. J. Miller, Tetrahedron 2009, 65, 4069; V. Declerck, J. Martinez, F. Lamaty, Chem. Rev. 2009, 109, 1; V. Singh, S. Batra, Tetrahedron 2008, 64, 4511; D. Basavaiah, K. V. Rao, R. J. Reddy, Chem. Soc. Rev. 2007, 36, 1581; D. Basavaiah, A. Jaganmohan Rao, T. Satyanarayana, Chem. Rev. 2003, 103, 811; P. Langer, Angew. Chem., Int. Ed. 2000, 39, 3049; E. Ciganek, in Organic Reactions, Ed. L. A. Paquette, Wiley, New York, 1997, Vol. 51, p. 201; D. Basavaiah, P. Dharma Rao, R. Suguna Hyma, Tetrahedron 1996, 52, 8001; S. E. Drewes, G. H. P. Roos, Tetrahedron 1988, 44, 4653.
- [5] A. Trofimov, V. Gevorgyan, Org. Lett. 2009, 11, 253; E. J. Lenardão, J. de Oliveiros Feijó, S. Thurow, G. Perin, R. G. Jacob, C. C. Silveira, Tetrahedron Lett. 2009, 50, 5215; N. Abermil, G. Masson, J. Zhu, J. Am. Chem. Soc. 2008, 130, 12596; M. Shi, X.-G. Liu, Org. Lett. 2008, 10, 1043; P. Shanmugam, V. Vaithiyanathan, K. Selvakumar, Tetrahedron Lett. 2008, 49, 2119; M. Bakthadoss, N. Sivakumar, G. Sivakumar, G. Murugan, Tetrahedron Lett. 2008, 49, 820; F. C. Pigge, R. Dhanya, E. R. Hefgen, Angew. Chem., Int. Ed. 2007, 46, 2887; S. Chuprakov, D. A. Malyshev, A. Trofimov, V. Gevorgyan, J. Am. Chem. Soc. 2007, 129, 14868; D. Basavaiah, K. Ramesh Reddy, N. Kumaragurubaran, Nat. Protoc. 2007, 2, 2665; A. Kattuboina, P. Kaur, C. Timmons, G. Li, Org. Lett. 2006, 8, 2771; M. E. Krafft, J. A. Wright, Chem. Commun. 2006, 2977; L. Navarre, S. Darses, J.-P. Genet, Adv. Synth. Catal. 2006, 348, 317; V. Singh, G. P. Yadav, P. R. Maulik, S. Batra, Tetrahedron 2006, 62, 8731; G. W. Kabalka, B. Venkataiah, C. Chen, Tetrahedron Lett. 2006, 47, 4187; J. Srivardhana Rao, J.-F. Brière, P. Metzner, D. Basavaiah, Tetrahedron Lett. 2006, 47, 3553; M. Dadwal, R. Mohan, D. Panda, S. M. Mobin, I. N. N. Namboothiri, Chem.Commun. 2006, 338; S. C. Kim, S. Gowrisankar, J. N. Kim, Tetrahedron Lett. 2006, 47, 3463; P. V. Ramachandran, S. Madhi, L. Bland-Berry, M. V. R. Reddy, M. J. O'Donnell, J. Am. Chem. Soc. 2005, 127, 13450; M. Shi, L.-H. Chen, C.-Q. Li, J. Am. Chem. Soc. 2005, 127, 3790; K. Matsui, S. Takizawa, H. Sasai, J. Am. Chem. Soc. 2005, 127, 3680; T. Turki, J. Villiéras, H. Amri, Tetrahedron Lett. 2005, 46, 3071; L. S. Santos, C. H. Pavam, W. P. Almeida, F. Coelho, M. N. Eberlin, Angew. Chem., Int. Ed. 2004, 43, 4330; D. Basavaiah, A. K. D. Bhavani, S. Pandiaraju, P. K. S. Sarma, Synlett 1995, 243; D. Basavaiah, S. Pandiaraju, P. K. S. Sarma, Tetrahedron Lett. 1994, 35, 4227; D. Basavaiah, V. V. L. Gowriswari, P. K. S. Sarma, P. Dharma Rao, Tetrahedron Lett. 1990, 31, 1621.
- [6] A. M. M. Castro, Chem. Rev. 2004, 104, 2939; P. Wipf, in 'Comprehensive Organic Synthesis', Eds. B. M. Trost, I. Fleming, L. Paquette, Pergamon Press, New York, 1990, Vol. 5, p. 827; W. S. Johnson, L. Werthemann, W. R. Bartlett, T. J. Brocksom, T.-T. Li, D. J. Faulkner, M. R. Petersen, J. Am. Chem. Soc. 1970, 92, 741.
- [7] a) D. Basavaiah, S. Pandiaraju, Tetrahedron Lett. 1995, 36, 757; b) D. Basavaiah, S. Pandiaraju, M. Krishnamacharyulu, Synlett 1996, 747; c) B. Das, A. Majhi, J. Banerjee, Tetrahedron Lett. 2006, 47, 7619.
- [8] J. O. Osby, S. W. Heinzman, B. Ganem, J. Am. Chem. Soc. 1986, 108, 67; S. W. Heinzman, B. Ganem, J. Am. Chem. Soc. 1982, 104, 6801; T. Satoh, S. Suzuki, Y. Suzuki, Y. Miyaji, Z. Imai, Tetrahedron Lett. 1969, 10, 4555.
- [9] D. Basavaiah, V. V. L. Gowriswari, Synth.Commun. 1987, 17, 587.

Received October 5, 2009